

SOLUBILITÉ DES SELS ET DES HYDROXYDES à 25°C

↓ précipité
 ● soluble dans l'eau, dissocié
 X ce composé n'existe pas

	F ⁻	Cl ⁻	Br ⁻	I ⁻	S ²⁻	OH ⁻	CO ₃ ²⁻	NO ₂ ⁻	NO ₃ ⁻	PO ₄ ³⁻	SO ₄ ²⁻	CH ₃ COO ⁻
Na ⁺	●	●	●	●	●	●	●	●	●	●	●	●
K ⁺	●	●	●	●	●	●	●	●	●	●	●	●
NH ₄ ⁺	●	●	●	●	●	●	●	●	●	●	●	●
Ag ⁺	●	↓	↓	↓	↓	↓	↓	●	●	↓	●	●
Mg ²⁺	↓	●	●	●	●	↓	↓	●	●	↓	●	●
Ca ²⁺	↓	●	●	●	●	●	↓	●	●	↓	●	●
Ba ²⁺	●	●	●	●	●	●	↓	●	●	↓	↓	●
Sr ²⁺	↓	●	●	●	●	●	↓	●	●	↓	↓	●
Cd ²⁺	●	●	●	●	↓	↓	↓	●	●	↓	●	●
Co ²⁺	●	●	●	●	↓	↓	↓	●	●	↓	●	●
Cu ²⁺	●	●	●	X	↓	↓	↓	●	●	↓	●	●
Hg ²⁺	X	●	●	↓	↓	↓	↓	●	●	↓	X	●
Mn ²⁺	●	●	●	●	↓	↓	↓	●	●	↓	●	●
Ni ²⁺	●	●	●	●	↓	↓	↓	●	●	↓	●	●
Pb ²⁺	↓	●	●	↓	↓	↓	↓	●	●	↓	↓	●
Zn ²⁺	●	●	●	●	↓	↓	↓	●	●	↓	●	●
Fe ²⁺	↓	●	●	●	↓	↓	↓	X	●	↓	●	●
Fe ³⁺	●	●	●	●	↓	↓	X	X	●	↓	●	●
Sn ²⁺	●	●	●	●	↓	↓	↓	X	X	↓	●	X
Sn ⁴⁺	●	●	●	●	↓	↓	X	X	X	↓	●	●
Al ³⁺	●	●	●	●	X	↓	X	X	●	↓	●	●
Bi ³⁺	↓	X	X	↓	↓	↓	X	X	X	↓	X	↓
Cr ³⁺	↓	●	●	●	↓	↓	X	X	●	↓	●	●