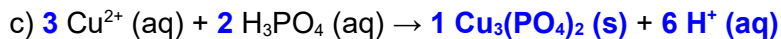
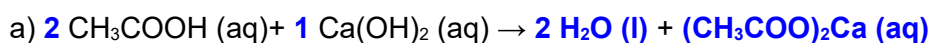
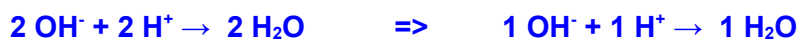
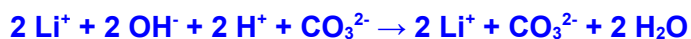
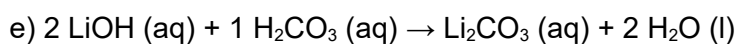
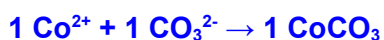
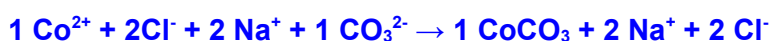
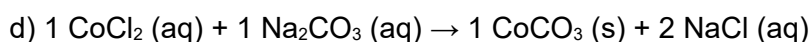


Nom de l'anion		Formule du cation		Nom de l'acide
sulfate	SO_4^{2-}	2H^+	H_2SO_4	acide sulfurique
nitrite	NO_2^-	H^+	HNO_2	acide nitreux
sulfure	S^{2-}	2H^+	H_2S	acide sulfhydrique
acétate	CH_3COO^-	H^+	CH_3COOH	acide acétique

Donnez les équations globales des réactions ioniques suivantes :

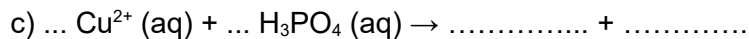
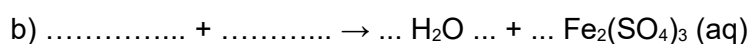
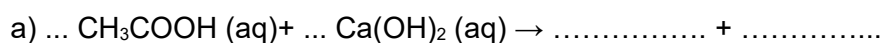


Pour la réactions ioniques suivantes, donnez les équations ioniques et ioniques réduites.

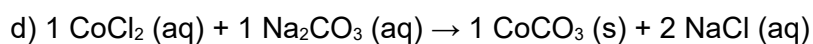


Nom de l'anion		Formule du cation		Nom de l'acide
sulfate	SO_4^{2-}	2H^+	H_2SO_4	acide sulfurique
	NO_2^-			
				acide sulfhydrique
			CH_3COOH	

Donnez les équations globales des réactions ioniques suivantes :

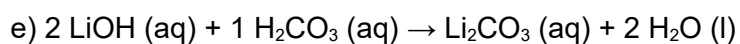


Pour la réactions ioniques suivantes, donnez les équations ioniques et ioniques réduites.



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